Simple Synthesis of (±)-(E)-3-(4-Hydroxyphenyl)-N-[4-(3-methyl-2,5dioxo-1-pyrrolidinyl)butyl]-2-propenamide, a Novel Phenolic Amide Derivative from the Bulbs of *Lilium regale* WILSON

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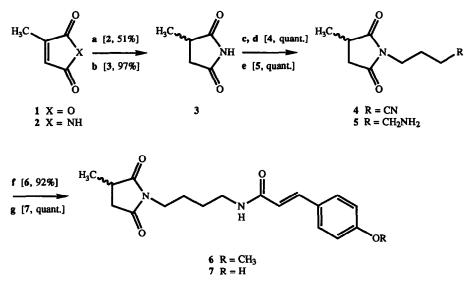
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Abstract: A synthesis of (\pm) -(E)-3-(4-hydroxyphenyl)-N-[4-(3-methyl-2,5-dioxo-1-pyrrolidinyl)butyl]-2-propenamide (7) is described. This phenolic amide was prepared in six steps with an overall yield of 46%.

A reinvestigation of the methanolic extract of *Lilium regale* WILSON, undertaken by *Mimaki* and *Sashida*, led to isolation and characterization of the novel phenolic amide 7.¹ Known from various plants, conjugates of hydroxycinnamic acid with aliphatic amines such as putrescine, spermidine and spermine are especially accumulated in their reproductive organs.² Some of the amides are suggested to protect the plants from viral, bacterial or fungal infection and may be produced as phytoalexins.³ *L. regale* is known for its strong resistance to viral diseases.⁴

As we have a continuing interest in compounds of natural origin, especially the polyamine alkaloids,⁵ we were prompted to investigate the synthesis and spectroscopic behavior of the putrescine alkaloid (\pm) -(*E*)-3-(4-hydroxyphenyl)-*N*-[4-(3-methyl-2,5-dioxo-1-pyrrolidinyl)butyl]-2-propenamide (7). In this paper, we describe a facile preparation of this phenolic amide, the structure of which was established by spectroscopic analysis and by comparison with published data of the natural product.

The synthesis of (\pm) -(E)-3-(4-hydroxyphenyl)-N-[4-(3-methyl-2,5-dioxo-1-pyrrolidinyl)butyl]-2propenamide (7), as shown, begins by preparing citraconimide (2) using citraconic anhydride (1) and heating it under reflux in the presence of NH₄OAc/AcOH. This method to prepare the imide 2 was reported by *Earl et al.*,⁶ but by a slight modification a higher yield was achieved (*vide infra*). The reaction proceeds through the 2methylmaleamic acid intermediate to give 2 and substantial quantities of resinous byproducts presumably of the general formula -[NRC(O)CH=CHC(O)]_x-.⁷



a) NH₄OAc, AcOH b) H₂, Pd/C, EtOH c) Na/MeOH d) Br(CH₂)₃CN, DMF e) H₂, PtO₂, EtOH/HCl f) 4-methoxycinnamic acid, 1-methyl-2-chloromethylpyridinium iodide, Et₃N, CH₂Cl₂ g) BBr₃, CH₂Cl₂

Scheme

In the next step, 2 was hydrogenated in the presence of 10% Pd/C in EtOH. The few percent of imide 3 which had openned could easily be removed by chromatography. Reduction at this stage was very important because of polymerization found to occur when alkylation was directly attempted with 2.

Alkylation was accomplished then by addition of 4-bromobutyronitrile to the dry sodium salt of 3 in absolute DMF.⁸ After heating under reflux, the DMF was evaporated and the residue treated with CH₂Cl₂ to deliver after further purification 4 in a quantitative yield.

The transformation of the nitrile group in 4 to the corresponding primary amine by catalytic hydrogenation was seen as the best method for reduction.⁹ In fact, (\pm) -N-(4-aminobutyl)-3-methylsuccinimide (5) was quantitatively won in form of its HCl salt by treatment of 4 with H₂ (ca. 3 atm) and PtO₂ in ethanolic HCl.

Following, the carboxamide 6 could be prepared in 92% yield when equimolar amounts of the primary amine 5 and 4-methoxycinnamic acid were treated with 1.2 molar amounts of *Mukaiyama's* reagent¹⁰ (1-methyl-2-chloropyridinium iodide) in the presence of 2.4 molar amounts of Et₃N.

Finally, the demethylation of the carboxamide 6 proved not to be a trivial affair. For deprotection the use of NaCN/DMSO and iodotrimethylsilane were tried, but both methods failed. The demethylation was carried out with BBr₃ which delivered the product 7 in quantitative yield. This preparation gave (\pm) -(E)-3-(4-hydroxyphenyl)-N-[4-(3-methyl-2,5-dioxo-1-pyrrolidinyl)butyl]-2propenamide (7) in 46% overall yield. The first step, the preparation of the imide, remains problematic though because of its elusiveness to a simple high yield production.6,7,11

EXPERIMENTAL

General. All chemicals used were of high commercial quality. Solvents used for chromatography were distilled as usual prior to use. Melting points were determined on a *Mettler FP-5* instrument. UV, in nm (log e), and IR spectra were measured on a *Perkin-Elmer 555* and *Perkin-Elmer 781* spectrophotometers respectively. ¹H-NMR spectra were carried out on either on a *Bruker AC-300* or *AM-400* spectrometer; ¹³C-NMR (at 50.4 MHz) on a *Varian XL-200*. Chemical shifts are reported in ppm (δ scale) with CDCl₃, unless otherwise stated, as internal standard. EI-MS (at 70 eV) or CI-MS (NH₃ as reactant gas) were measured either on a *Varian MAT 112S* or *Finnigan MAT TSQ 700* mass spectrometer. Intensities for EI are only given for values \geq 10%, except for *M*^{+*}, from m/z \geq 40. For flash chromatography, *Merck* silica gel 60 (0.04-0.06 mm) was used. TLC were done on precoated Kieselgel 60 *F*₂₅₄ aluminum plates (*Merck*); spots visualized under UV light (254 nm) and by staining reagents.

Citraconimide (2). NH₄OAc (35 g, 454 mmol, predried for 2 h/3 x 10^{-2} Torr) and citraconic anhydride (1, 25 ml, 179 mmol) were added to AcOH (50 ml) and heated under reflux. After 2 h, the solution was cooled to room temperature and evaporated *in vacuo* at 70°. Ice water (200 ml) was added to the dark syrupy residue and this was followed by extraction with EtOAc (8 x 25 ml) and CH₂Cl₂ (2 x 25 ml). The combined organic extracts were evaporated to give a yellow solid which was then bulb-to-bulb distilled; the fraction distilling at 120-130° (air bath temperature)/3 x 10^{-2} Torr was collected as a colorless solid (10.93 g, 55%). Subsequent purification by flash chromatography (EtOAc/hexane 1:1) afforded 2 (10.06 g, 51%). M.p. (toluene): 104.5-106° (lit.,⁶ 103.5-105.5°). IR (KBr): 3380s, 3250m, 3080m, 2985m, 2940m, 1690s + 1645s (imide-(CO)), 1590m, 1545s, 1535s, 1520s, 1465m, 1430m, 1370s, 1330m, 1305m, 1275s, 1250m, 1205m, 1165s, 1125m, 1075w, 1040w, 1020w, 1010w, 955m, 900w, 885w, 875w, 810w, 795w, 775w, 760w, 735w, 690w, 625w. ¹H-NMR (400 MHz, (d₆)-DMSO): 10.68 (br. s, NH, exchangeable); 6.50-6.45 (m, H-C(4)); 1.93 (d, J = 1.8, CH₃). ¹H-NMR (400 MHz): 7.27 (br. s, NH); 6.35-6.30 (m, H-C(4)); 2.08 (d, J = 1.8, CH₃). ¹³C-NMR: 171.7, 170.6 (2 s, CO); 146.7 (s); 128.3 (d); 10.8 (q). CI-MS: 223 (100, [2M + 1]⁺), 112 (89, [M + 1]⁺). Anal. calc. for C₅H₅NO₂ (111.10): C 54.06, H 4.54, N 12.61; found: C 53.96, H 4.52, N 12.69.

(\pm)-3-Methylsuccinimide (3). The imide 2 (2.5 g, 22.5 mmol) was dissolved in EtOH (25 ml) containing a suspension of 10% Pd/C (125 mg) and stirred overnight under a blanket of H₂ (1 atm). The solution was filtered through cotton wool and evaporated to dryness yielding 3 as an oil, which on standing slowly crystallized. The purified 3 (2.43 g, 97%) was obtained as a colorless solid by flash chromatography (EtOAc/hexane 1:1). M.p. (toluene): 63.5-64.5° (lit.,¹² 62°). IR (KBr): 3500 (br.), 3060m, 2980m, 2940w, 2880w, 2760w, 1765s + 1710s (imide-(CO)), 1560w, 1540w, 1455w, 1415w, 1380m, 1350m, 1315m, 1290m, 1250w, 1205s, 1180s, 1120w, 1085w, 1035m, 930w, 890w, 795m, 725w, 640w. ¹H-NMR (400 MHz): 8.70 (br. s, NH); 3.00-2.85 (m, 2 H-C(4)); 2.45-2.30 (m, H-C(3)); 1.35 (d, J = 7.1, CH₃). ¹³C-NMR: 181.1, 176.8 (2 s, CO); 37.6 (t); 36.1 (d); 16.5 (q). EI-MS: 113 (39, M^{++}), 70 (32), 44 (11), 43 (12), 42 (100), 41 (34). Anal. calc. for C₅H₇NO₂ (113.12): C 53.09, H 6.24, N 12.38; found: C 53.15, H 6.26, N 12.26.

(±)-N-(3-Cyanopropy)-3-methylsuccinimide (4). Sodium (230 mg, 10 mmol) was firstly placed in MeOH (25 ml) followed by addition of imide 3 (1.13 g, 10 mmol). After complete dissolution, the solvent was evaporated to give a colorless crystalline mass. DMF (25 ml) and 4-bromobutyronitrile (1.98 ml, 20 mmol) were then added and the mixture was heated under reflux while stirring for 1 h. The DMF was removed *in vacuo* and the residue treated with CH₂Cl₂ (25 ml) from which the precipitate was filtered off. The filtrate was then washed (H₂O), dried (MgSO₄), and evaporated and the residual oil purified by flash chromatography (EtOAc/hexane 1:1) delivering 4 (1.80 g, quant.) as a colorless oil. IR (Film): 3600w, 3460w, 2980w, 2940m, 2880w, 2250w (CN), 1775m + 1705s (imide-(CO)), 1440m, 1405s, 1375m, 1360m, 1290w, 1265w, 1250m, 1220w, 1200w, 1195w, 1165s, 1120w, 1090w, 1070w, 1020m, 920w, 890w, 840w, 780w, 750w, 690w. ¹H-NMR (300 MHz): 3.63 (t, J = 7.0, NCH₂); 3.00-2.85 (m, 2 H); 2.40-2.30 (m, 1H); 2.36 (t, J = 7.0, CH₂CN); 1.35 (d, J = 7.2, CH₃). ¹³C-NMR: 180.5, 176.5 (2 s, CO); 118.9 (s, CN); 37.6, 36.3 (2 t); 34.6 (d); 23.5 (t); 16.4 (q); 15.1 (t). EI-MS: 180 (42, M^{+*}), 127 (56), 114 (74), 111 (48), 73 (60), 68 (47), 54 (43), 42 (81), 41 (100). Anal. calc. for C₉H₁₂N₂O₂ (180.21): C 60.00, H 6.71, N 15.54; found: C 60.14, H 6.90, N 15.28.

(\pm)-N-(4-Aminobutyl)-3-methylsuccinimide (5). To a solution of concentrated aqueous HCl (6.2 ml, 64.4 mmol) in EtOH (100 ml) was added PtO₂ (325 mg). This catalyst was firstly activated by agitation for 1 h under a blanket of H₂ (3 atm). Addition of 4 (4.64 g, 25.7 mmol) followed, and the mixture was further shaken for 8 h under H₂. The mixture was filtered then through cotton wool and evaporated yielding 5•HCl (5.70 g, quant.) as a colorless oil. IR (CHCl₃): 3450 (br.), 3035 (br., NH₃+), 2970 (br.), 1770m + 1700s (imide-(CO)), 1610m + 1515m (NH₃+), 1455m, 1440m, 1405s, 1365m, 1345m, 1290m, 1235m, 1145m, 1095w, 1050w, 1015w, 915w, 890w, 710w, 690w, 660w. ¹H-NMR (300 MHz, DMSO): 7.91 (br. s, NH₃+); 3.40-3.30 (m, 2 H); 2.90-2.70 (m, 4 H); 2.35-2.25 (m, 1 H); 1.60-1.45 (m, 4 H); 1.19 (d, J = 6.8, CH₃). CI-MS: 369 (100, [(2M + 1)]+), 185 (86, [M + 1]+).

(\pm)-(E)-3-(4-Methoxyphenyl)-N-[4-(3-methyl-2,5-dioxo-1-pyrrolidinyl)butyl]-2-propenamide (6). To a suspension of 1-methyl-2-chloropyridinium iodide (3.07 g, 12 mmol) in CH₂Cl₂ (100 ml) was added 4-methoxycinnamic acid (1.78 g, 10 mmol) and Et₃N (3.35 ml, 24 mmol) under a N₂ atmosphere. The addition of 5 (2.43 g, 11 mmol), dissolved in a solution of CH₂Cl₂ (10 ml) and Et₃N (1.5 ml, 11 mmol), was then added dropwise to the mixture while stirring. Afterwards, the mixture was heated under reflux and for 1 h. Upon cooling to room temperature, Et₂O (200 ml) was added and the resulting mixture was washed with a 5% aqueous HCl solution (3 x 100 ml) and with H₂O (1 x 100 ml). The organic layer was then evaporated and the residue, purified by flash chromatography (EtOAc), yielded the carboxamide 6 (3.16 g, 92%) as a colorless solid. M.p. (EtOAc): 122.5-123.5°. IR (CHCl₃): 3450w (NH), 3000m, 2940w, 2870w, 2840w, 1765m + 1700s (imide-(CO)), 1625m (amide-(CO)), 1605m, 1575w, 1515s (arom. ring), 1460w, 1440m, 1420w, 1400m, 1370m, 1340w, 1305w, 1285m, 1255m, 1195w, 1175s, 1130w, 1030m, 980w, 825m (arom. ring). ¹H-NMR (300 MHz): 7.55 (AXd, J = 15.6, 1 H); 7.42 (d, J = 8.7, 2 arom. H); 6.86 (d, J = 8.7, 2 arom. H); 6.27 (AXd, J = 15.6, 1)

1 H); 5.98 (br. s, NH); 3.81 (s, OCH₃); 3.51 (t, J = 7.0, 2 H); 3.44-3.35 (m, 2 H); 2.95-2.77 (m, 2 H); 2.33-2.26 (m, 1 H); 1.69-1.50 (m, 4 H); 1.32 (d, $J = 7.1, CH_3$). ¹³C-NMR: 180.4, 176.3, 166.3 (3 s, CO); 160.5 (s, arom. C); 139.9 (d); 129.0 (d, 2 arom. C); 127.4 (s, arom. C); 118.5 (d); 113.5 (d, 2 arom. C); 55.1 (q, OCH₃); 38.9, 38.0, 36.1 (3 t); 34.4 (d); 26.6, 25.0 (2 t); 16.5 (q). EI-MS: 344 (3, M^{+*}), 205 (24), 177 (26), 176 (14), 162 (14), 161 (100), 133 (27). CI-MS: 346 (23, $[M + 2]^+$), 345 (100, $[M + 1]^+$). Anal. calc. for C₁₉H₂₄N₂O₄ (344.41): C 66.26, H 7.02, N 8.13; found: C 66.14, H 7.20, N 8.32.

(±)-(E)-3-(4-Hydroxyphenyl)-N-[4-(3-methyl-2,5-dioxo-1-pyrrolidinyl)butyl]-2-propenamide (7). Under a N₂ atmosphere, a solution of 6 (100 mg, 0.29 mmol) in CH₂Cl₂ (2.5 ml) was cooled in a dry ice/2-propanol bath to which BBr3 (0.09 ml, 0.96 mmol) was introduced using a syringe. The bath was then removed and the mixture was stirred for 30 min, poured into ice water, stirred for a further 30 min, saturated with salt and finally extracted with CH₂Cl₂. The extract was dried (MgSO₄) and the solvent removed, yielding 7 (96 mg, quant.) as a colorless solid. M.p. (MeOH): 192.5-193.0°. UV (EtOH): 308 (4.36), 292 (4.37), 225 (4.17), 203 (4.35). UV (ca. 1.3 x 10⁻³M NaOH/EtOH soln.): 353, 312 sh, 235. IR (KBr): 3450m, 3350s (NH), 3125s (OH), 3020m, 2960m, 2875m, 2820m, 2780w, 2700w, 2630w, 1900w, 1770m + 1695s (five-membered cyclic inide), 1655s (sec. amide, A band, C=O), 1605s, 1585s, 1520s (arom. ring), 1555s, (sec. amide, B band, NH), 1455s, 1410s, 1375s, 1350s, 1315s, 1285s, 1250s, 1230s, 1205m, 1175s, 1140m, 1100m, 1085m, 1035w, 995m, 965w, 955w, 920w, 905w, 885w, 860w, 840s, 795w, 750m, 715w, 655w, 635w. ¹H-NMR (400 MHz, C₅D₆N): 11.98 (br. s, OH); 8.50 (br. t, J = 5.5, NH); 8.05 (AXd, J = 15.6, 1 H); 7.50 (d, J = 8.5, 2 H); 7.08 (d, J = 8.5, 2 H); 6.76 $(AXd, J = 15.6, 1 \text{ H}); 3.60-3.50 \text{ (m, 4 H)}; 2.83 \text{ (dd, } J = 16.6, 9.0, 1 \text{ H}); 2.85-2.75 \text{ (m, 1 H)}; 2.27 \text{ (dd, } J = 13.1, 1.1); 3.60-3.50 \text{ (m, 4 H)}; 2.83 \text{ (dd, } J = 16.6, 9.0, 1 \text{ H}); 3.60-3.50 \text{ (m, 4 H)}; 3.60-3.50 \text$ 3.6, 1 H); 1.67 (m, 4 H); 1,18 (d, J = 7.1, CH₃). ¹³C-NMR (C₅D₆N): 181.0, 176.8, 166.8 (3 s, CO); 160.6 (s, arom. C); 140.2 (d); 130.1 (d, 2 arom. C); 127.2 (s, arom. C); 119.6 (d); 116.9 (d, 2 arom. C); 39.5, 38.6, 36.7 (3 t); 35.1 (d); 27.7(t); 25.8 (t); 16.5 (q). EI-MS: 330 (6, M⁺⁺), 204 (10), 183 (23), 162 (12), 148 (13), 147 (100), 119 (19), 91 (19), 70 (19), 69 (17), 65 (11), 57 (33), 56 (15), 55 (21). CI-MS: 331 (83, [M + 1]⁺). Anal. calc. for C₁₈H₂₂N₂O₄ (330.39); C 65.44, H 6.71, N 8.48; found: C 65.73, H 6.57, N 8.23.

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